



Pergamon

Tetrahedron Letters 41 (2000) 759–762

TETRAHEDRON
LETTERS

A novel approach to conformationally restricted analogues of nicotine and anabasine by an intramolecular Hamaguchi–Ibata reaction

Tarun K. Sarkar,* Sankar Basak and Sunil K. Ghosh

Department of Chemistry, Indian Institute of Technology, Kharagpur 721302, India

Received 8 September 1999; revised 12 November 1999; accepted 17 November 1999

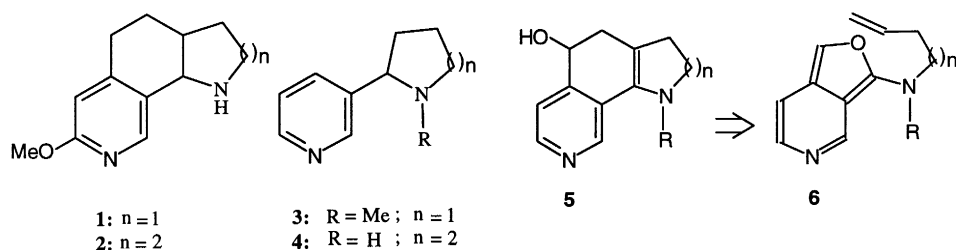
Abstract

A short synthesis of methyl (*SR*)-8,10-dichloro-1,2,3,4,5,6-hexahydro-6-hydroxy-1-methyl-1,9-phenanthroline-6-carboxylate (**14**) has been achieved en route to conformationally restricted analogues of nicotine and anabasine. The key feature of this process involves an efficient intramolecular Hamaguchi–Ibata reaction of **11**. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: azaisobenzofuran; intramolecular cycloaddition; domino reaction; 1,9-phenanthroline.

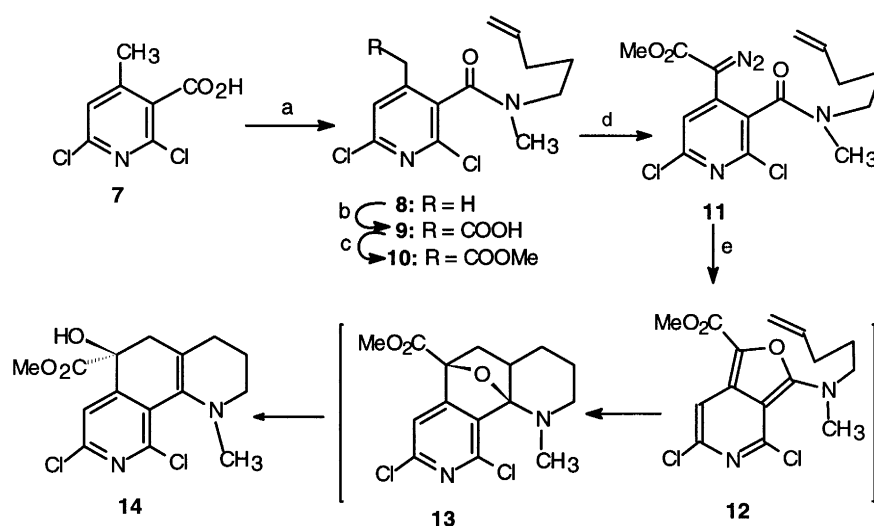
The examination of the role of particular conformations on the pharmacological properties of important biologically active molecules has remained an active area of investigation. One approach to such a study is to modify the parent molecule in such a fashion that its original conformational mobility is severely limited to one particular conformation.¹ In light of this, recently a series of conformationally restricted analogues, e.g. **1** and **2** of nicotine (**3**) and anabasine (**4**) have been synthesized and evaluated as agonists of neuronal acetylcholine receptors (nAChRs).² In particular, compound **1** which selectively activates human recombinant $\alpha 2\beta 4$ and $\alpha 4\beta 4$ nAChRs has been shown to be active in animal models of Parkinson's disease and pain. The long and circuitous synthetic route² described for the ring systems **1** and **2** does not appear to be viable for the synthesis of a large variety of bridged nictines and anabasines required for biological testing. Hence, the development of an alternative and more flexible route which overcomes these limitations was felt desirable.

* Corresponding author.



We have recently reported on a Hamaguchi–Ibata reaction involving a stable azaisobenzofuran to give functionalized isoquinolines.³ In this letter, we describe for the first time an intramolecular⁴ version **6**→**5** of this methodology which has allowed synthesis of **14**, a molecule which incorporates the basic ring system present in bridged anabasines.

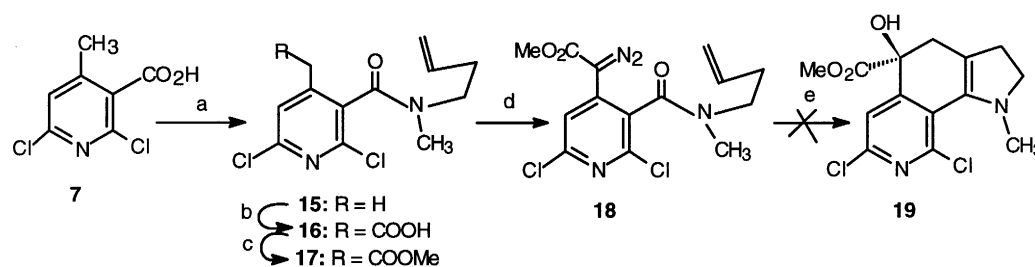
The substituted diazoacetic ester **11** (IR: 2110,1715,1637 cm^{-1}), the substrate for the intramolecular Hamaguchi–Ibata reaction, was made from the readily available carboxylic acid **7**³ by a standard synthetic protocol as shown in Scheme 1.⁵ When **11** was exposed to 1 mol% $\text{Rh}_2(\text{OAc})_4$ in refluxing benzene for 1 h, the bridged anabasine **14** (mp 165–167°C) was obtained in 52% yield as a yellow crystalline solid via intramolecular cycloaddition **12**→**13** followed by ring opening of **13** and subsequent proton transfer. The structure of **14** is supported by its elemental analysis, ^1H and ^{13}C NMR spectra⁶ (Scheme 1).



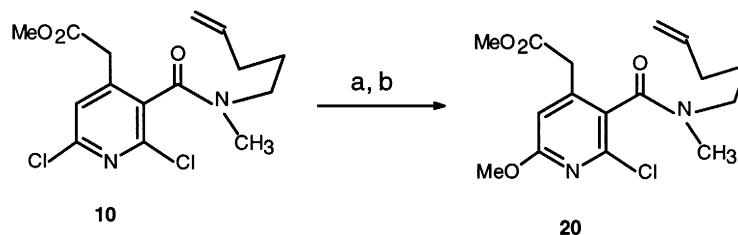
Scheme 1. (a) $(\text{COCl})_2$, PhH, reflux and then *N*-methylpent-4-enylamine, py, rt, 72%; (b) LDA, THF, -78°C and then CO_2 , 83%; (c) CH_2N_2 , 94%; (d) 4-acetamidobenzenesulfonyl azide, Et_3N , 0°C to rt, 12 h (Ref. 7), 93%; (e) $\text{Rh}_2(\text{OAc})_4$, PhH, reflux, 1 h, 52%

Unfortunately, when the dienophile tether was shortened by one methylene unit, no intramolecular cycloaddition leading to a bridged nicotine analogue **19** was obtained (Scheme 2).⁵ This difficulty may be overcome as recently shown by Padwa et al⁸ by replacing the amine tether (cf. **6**) by an amide tether.⁹

A noteworthy feature of our approach is that one of the chloro substituents flanking the nitrogen atom in **10** can be selectively replaced by a nucleophile, if desired. It is evident that there is considerable steric hindrance to nucleophilic attack at C-2 in **10**. Indeed, exposure of **10**^{5,10} to aqueous alkali in methanol and a follow-up treatment with diazomethane gave practically a single product to which structure **20**^{5,11} is assigned based on ^1H NMR analysis¹² (Scheme 3).



Scheme 2. (a) $(\text{COCl})_2$, PhH, reflux and then *N*-methylbut-3-enylamine, py, rt, 66%; (b) LDA, THF, -78°C and then CO_2 , 80%; (c) CH_2N_2 , 92%; (d) 4-acetamidobenzenesulfonyl azide, Et_3N , 0°C to rt, 12 h (Ref. 7), 89%; (e) $\text{Rh}_2(\text{OAc})_4$, PhH, reflux, 1 h



Scheme 3. (a) 10% aq. KOH, MeOH, 12 h, rt; H_3O^+ ; (b) CH_2N_2 (73% from **10**)

In conclusion, the domino cascade sequence¹³ presented herein has allowed ready access to a bridged anabasine analogue. Further work on bridged nicotines and their biological activities are in progress and will be reported in due course.

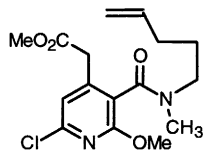
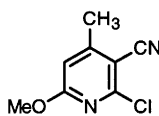
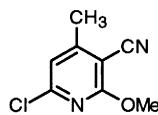
Acknowledgements

Financial support from CSIR & DST, Government of India is gratefully acknowledged. We are also grateful to Professor T. Gallagher (Bristol, UK), Dr. S. Djuric (Abbott, USA), Dr. M. K. Ghosh (Labline, Germany) and Dr. A. Sarkar (NCL, Pune) for help.

References

- Chavdarian, C. G.; Seeman, J. I.; Wooten, J. B. *J. Org. Chem.* **1983**, *48*, 492 and references cited therein.
- Vernier, J.-M.; Holsenback, H.; Cosford, N. D. P.; Whitten, J. P.; Menzaghi, F.; Reid, R.; Rao, T. S.; Saccaan, A. I.; Lloyd, G. K.; Suto, C. M.; Chavez-Noriega, L. E.; Washburn, M. S.; Urrutia, A.; McDonald, I. A. *Bioorg. Med. Chem. Lett.* **1998**, *8*, 2173.
- Sarkar, T. K.; Ghosh, S. K.; Nandy, S. K.; Chow, T. J. *Tetrahedron Lett.* **1999**, *40*, 397.
- Peters, O.; Friedrichsen, W. *Tetrahedron Lett.* **1995**, *36*, 8581.
- Compounds **8–11**, **15–18** and **20** exist as mixtures of rotamers in solution.
- IR (KBr) ν_{max} : 3378, 1745, 1616 cm^{-1} ; ^1H NMR (200 MHz, CDCl_3): δ 7.41 (s, 1H), 3.69 (s, 3H), 3.20 (bs, 1H), 3.19–3.09 (m, 1H), 2.90–2.75 (m, 1H), 2.65 (1H, dt, $J=15$ and 1.8 Hz), 2.44 (1H, d, $J=15.1$ Hz), 2.40 (s, 3H), 2.20–2.10 (m, 2H), 1.80–1.65 (m, 2H); ^{13}C NMR (50 MHz, CDCl_3): δ 173.4, 152.6, 147.7, 145.1, 138.2, 125.9, 118.5, 113.7, 74.9, 52.9, 51.3, 41.2, 40.8, 28.8, 18.5. Anal. calcd for $\text{C}_{15}\text{H}_{16}\text{N}_2\text{O}_3\text{Cl}_2$: C, 52.49; H, 4.69; N, 8.16; found: C, 52.73; H, 4.62; N, 8.10.
- Baum, J. S.; Shook, D. A.; Davies, H. M. L.; Smith, H. D. *Synth. Commun.* **1987**, *17*, 1709.
- Padwa, A.; Kappe, C. O.; Cochran, J. E.; Snyder, J. P. *J. Org. Chem.* **1997**, *62*, 2786.
- This possibility is under investigation.
- ^1H NMR (300 MHz, CDCl_3): δ 7.34 and 7.28 (s, 1H), 5.91–5.78 and 5.71–5.60 (m, 1H), 5.12–4.90 (m, 2H), 3.80 (d, 1H, $J=16.2$ Hz), 3.72 and 3.71 (s, 3H), 3.56 (d, 1H, $J=16.2$ Hz), 3.70–3.43 and 3.17–2.96 (m, 2H), 3.11 and 2.87 (s, 3H), 2.20–1.66 (m, 4H).

11. ^1H NMR (300 MHz, CDCl_3): δ 6.69 and 6.64 (s, 1H), 5.93–5.77 and 5.76–5.60 (m, 1H), 5.11–4.90 (m, 2H), 3.95 and 3.94 (s, 3H), 3.77 (d, 1H, $J=16.2$ Hz), 3.70 and 3.68 (s, 3H), 3.51 (d, 1H, $J=16.2$ Hz), 3.72–3.42 and 3.17–3.01 (m, 2H), 3.09 and 2.88 (s, 3H), 2.35–1.65 (m, 4H).
12. Compound **20** is contaminated with a trace of its regioisomer **21**. In **20**, the C-5 proton resonates at δ 6.69 and 6.64 (two rotamers), whereas the C-5 proton in **21** resonates at δ 6.96 and 6.92 (two rotamers); all these signals have moved upfield in comparison to the C-5 signals (δ 7.34 and 7.28) in **10**. The chemical shifts of C-5 proton in the two rotamers of **20** match with the chemical shift of the C-5 proton in **22** (δ 6.6)¹⁴ and not **23** (δ 6.9).¹⁴

**21****22****23**

13. Ho, T.-L. *Tandem Organic Reactions*; Wiley-Interscience: New York, 1992; *Frontiers in Organic Synthesis, Chem. Rev.* **1996**, *96*, 1; Tietze, L. F.; Beifuss, U. *Angew. Chem., Int. Ed. Engl.* **1993**, *32*, 131; Padwa, A. *Chem. Commun.* **1998**, 1417.
14. Pelisson, M. M. M.; da Silva, G. V. J.; Clive, D. L. J.; Coltart, D. M.; Hof, F. A. *J. Heterocyclic Chem.* **1999**, *36*, 653.